

C(1)	0.2678 (3)	0.0693 (3)	0.5685 (3)	0.0413 (9)
C(2)	0.2500 (4)	0.2288 (3)	0.3278 (3)	0.0492 (11)
C(3)	0.2371 (4)	0.1159 (3)	0.2253 (4)	0.0527 (12)
C(4)	0.2356 (3)	-0.0200 (3)	0.2967 (3)	0.0465 (11)
C(5)	0.2479 (7)	0.3801 (4)	0.2702 (6)	0.0734 (17)
C(6)	0.2117 (6)	-0.1486 (4)	0.1893 (4)	0.0660 (16)
C(7)	0.2666 (3)	-0.3844 (3)	0.7365 (3)	0.0427 (9)

Table 2. Selected geometric parameters (\AA , $^\circ$)

S(1)—C(1)	1.685 (3)	S(2)—C(7)	1.705 (3)	
N(1)—C(1)	1.369 (4)	N(1)—C(2)	1.353 (4)	
N(2)—C(1)	1.344 (3)	N(2)—C(4)	1.337 (3)	
N(3)—C(7)	1.320 (3)	N(4)—C(7)	1.327 (3)	
C(2)—C(3)	1.354 (4)	C(2)—C(5)	1.493 (5)	
C(3)—C(4)	1.397 (4)	C(4)—C(6)	1.485 (5)	
C(1)—N(1)—C(2)	124.0 (2)	C(1)—N(2)—C(4)	119.4 (2)	
N(1)—C(1)—N(2)	118.4 (2)	S(1)—C(1)—N(2)	122.1 (2)	
S(1)—C(1)—N(1)	119.6 (2)	N(1)—C(2)—C(5)	117.2 (3)	
N(1)—C(2)—C(3)	117.2 (2)	C(3)—C(2)—C(5)	125.6 (3)	
C(2)—C(3)—C(4)	119.0 (3)	N(2)—C(4)—C(3)	122.0 (2)	
C(3)—C(4)—C(6)	122.0 (3)	N(2)—C(4)—C(6)	115.9 (2)	
N(3)—C(7)—N(4)	117.0 (2)	S(2)—C(7)—N(4)	122.1 (2)	
S(2)—C(7)—N(3)	121.0 (2)			
<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N(3)—H(31)...S(1)	0.87 (3)	2.58 (3)	3.418 (3)	164 (3)
N(4)—H(42)...N(2)	0.95 (3)	1.97 (3)	2.915 (4)	172 (3)
N(1)—H(1)...S(2 ⁱ)	0.89 (3)	2.40 (3)	3.277 (3)	167 (2)
N(3)—H(32)...S(2 ⁱⁱ)	0.90 (3)	2.60 (2)	3.462 (3)	161 (2)
N(4)—H(41)...S(2 ⁱⁱⁱ)	0.80 (3)	2.75 (2)	3.486 (3)	154 (3)

Symmetry codes: (i) $x, 1 + y, z$; (ii) $1 - x, -1 - y, 2 - z$; (iii) $-x, -1 - y, 1 - z$.

The thermal motion of the molecule is strongly anisotropic with U_{22} larger than the other components for most of the atoms.

Program systems used: XRAYARC (Vickery, Bright & Mallinson, 1971) for Patterson synthesis; NORMAL, EXFFT and SEARCH sequences of MULTAN78 (Main, Hull, Lessinger, Germain, Declercq & Woolfson, 1978) for weighted Fourier syntheses, SHELX76 (Sheldrick, 1976) for refinement, and ORTEPII (Johnson, 1976) for molecular graphics.

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates, bond distances and angles involving H atoms, least-squares-planes data and torsion angles have been deposited with the IUCr (Reference: MU1121). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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1-Diphenylmethylene-4-(triphenylmethyl)-cyclohexa-2,5-diene Benzene Solvate

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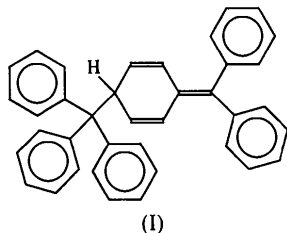
Abstract

The crystal structure of the benzene solvate of 1-diphenylmethylene-4-(triphenylmethyl)cyclohexa-2,5-diene, $C_{38}H_{30} \cdot C_6H_6$, has been determined by X-ray diffraction.

Comment

After the discovery of the Ph_3C^* radical there was lengthy discussion about the structure of its dimer (McBride, 1974). The correct structure of this dimer, 1-diphenylmethylene-4-(triphenylmethyl)cyclohexa-2,5-diene, (I), was first established by an NMR study (Staab, Brettschneider & Brunner, 1970) and was later confirmed by X-ray analysis of the ethyl acetate solvate and unsolvated crystals (Allemand & Gerdil, 1978;

Blom, Roelofsen & Kanters, 1982). In the course of our systematic investigation of the reaction of metallic lanthanides with organoelement halides of Group IVB elements, we found that the reaction of Yb with Ph₃CCl gave a precipitate of YbCl₂ and a yellow solution of Ph₃C• radicals in equilibrium with their dimer (Bochkarev, Molosnova, Zakharov, Fukin, Yanovsky & Struchkov, 1994). Compound (I) was isolated as crystals of its benzene solvate by crystallization from benzene/hexane solution. The crystals were stable in air for a few hours and were readily soluble in benzene and tetrahydrofuran, and sparingly soluble in hexane. They melted, with decomposition, at 413–417 K.



The structure of (I) found here is similar to that in both the ethyl acetate solvate (Allemand & Gerdil, 1978) and the unsolvated crystals (Blom, Roelofsen & Kanters, 1982). The C(20)–C(26) fragment of the molecule is planar to within 0.02 Å. The double bonds C(21)=C(22) [1.335 (4) Å] and C(24)=C(25) [1.337 (4) Å] are equal and are close to the average C=C bond length found for cyclohexa-1,3-dienes (1.332 Å; Allen, Kennard, Watson, Brammer, Orpen & Taylor, 1987). The C(26) atom has a planar trigonal environment; the C(23)=C(26), C(26)–C(27) and C(26)–C(33) bond lengths are 1.376 (4), 1.485 (4) and 1.483 (4) Å, respectively. The C(1)–C(20) bond length [1.597 (4) Å] is unusually long, just as it is in the ethyl acetate solvate structure (1.589 Å; Allemand & Gerdil, 1978) and in the two independent molecules of the unsolvated structure (1.60 and 1.61 Å; Blom, Roelofsen & Kanters, 1982). In these three structures, the differences between the bond lengths and angles are not greater than 3σ but there are, however, some conformational differences. In the benzene solvate structure, the dihedral angles between the phenyl rings C(27)–C(32) and C(33)–C(38), and the plane composed of atoms C(23), C(27) and C(33), are 51.68 (9) and 54.9 (1)°, respectively, and the dihedral angle between the average planes of the C(20)–C(25) and C(14)–C(19) fragments is 55.1 (1)°. The equivalent angles in the ethyl acetate solvate structure are 53.1, 55.2 and 52.2° (Allemand & Gerdil, 1978), and in the two independent molecules of the unsolvated structure the angles are 93.3, 62.6 and 41.6°, and 90.8, 45.9 and 39.4°, respectively (Blom, Roelofsen & Kanters, 1982). The packing in the ethyl acetate and benzene solvate structures is similar. The unit-cell lengths are similar but the differences in an-

gles are more significant (*cf.* $a = 13.690$, $b = 12.198$, $c = 10.776$ Å, $\alpha = 81.96$, $\beta = 79.06$ and $\gamma = 70.15^\circ$ for the ethyl acetate solvate crystal).

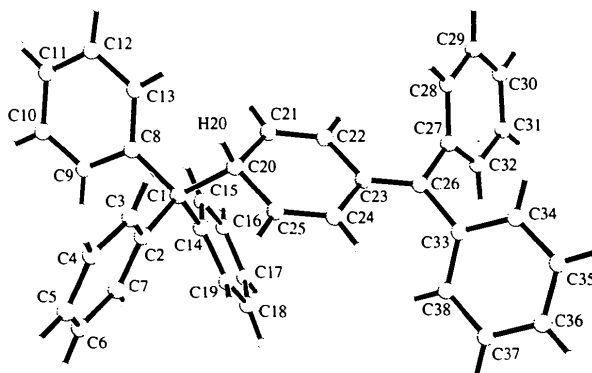


Fig. 1. A general view of 1-diphenylmethylene-4-(triphenylmethyl)-cyclohexa-2,5-diene with the labelling of non-H atoms shown.

Experimental

Crystal data

C₃₈H₃₀·C₆H₆
M_r = 564.73
 Triclinic
P $\bar{1}$
 $a = 10.555$ (3) Å
 $b = 12.680$ (3) Å
 $c = 13.675$ (4) Å
 $\alpha = 62.42$ (2)°
 $\beta = 87.01$ (2)°
 $\gamma = 85.63$ (2)°
 $V = 1617.2$ (8) Å³
 $Z = 2$
 $D_x = 1.160$ Mg m⁻³

Mo *K*α radiation
 $\lambda = 0.71073$ Å
 Cell parameters from 24 reflections
 $\theta = 12$ – 12.5°
 $\mu = 0.065$ mm⁻¹
 $T = 193$ (2) K
 Parallelepiped
 0.6 × 0.5 × 0.4 mm
 Light yellow

Data collection

Syntex P2₁ diffractometer
 $\theta/2\theta$ scans
 Absorption correction: none
 5197 measured reflections
 4882 independent reflections
 3251 observed reflections
 $[I > 2\sigma(I)]$
 $R_{int} = 0.0516$

$\theta_{max} = 24.05^\circ$
 $h = -2 \rightarrow 11$
 $k = -14 \rightarrow 14$
 $l = -15 \rightarrow 15$
 2 standard reflections monitored every 98 reflections
 intensity decay: 1.1%

Refinement

Refinement on F^2
 $R(F) = 0.0525$
 $wR(F^2) = 0.1790$
 $S = 1.038$
 4871 reflections
 541 parameters
 All H-atom parameters refined
 $w = 1/[\sigma^2(F_o^2) + (0.0976P)^2 + 0.4246P]$
 where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{max} = -0.199$
 $\Delta\rho_{max} = 0.269$ e Å⁻³
 $\Delta\rho_{min} = -0.280$ e Å⁻³
 Atomic scattering factors from *International Tables for Crystallography* (1992, Vol. C, Tables 4.2.6.8 and 6.1.1.4)

Table 1. *Fractional atomic coordinates and equivalent isotropic displacement parameters* (\AA^2)
$$U_{\text{eq}} = (1/3)\sum_i \sum_j U_{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j.$$

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{eq}
C1	-0.3106 (3)	-0.2543 (2)	0.0643 (2)	0.0326 (6)
C2	-0.1935 (3)	-0.2699 (2)	0.1337 (2)	0.0337 (6)
C3	-0.1846 (3)	-0.2002 (3)	0.1874 (2)	0.0431 (7)
C4	-0.0852 (3)	-0.2190 (3)	0.2567 (3)	0.0521 (9)
C5	0.0079 (3)	-0.3072 (3)	0.2754 (3)	0.0577 (10)
C6	0.0014 (3)	-0.3776 (3)	0.2244 (3)	0.0533 (9)
C7	-0.0984 (3)	-0.3604 (3)	0.1549 (2)	0.0445 (8)
C8	-0.4154 (3)	-0.3173 (2)	0.1507 (2)	0.0319 (6)
C9	-0.3991 (3)	-0.4406 (2)	0.2185 (2)	0.0377 (7)
C10	-0.4870 (3)	-0.5027 (3)	0.2990 (2)	0.0426 (8)
C11	-0.5950 (3)	-0.4434 (3)	0.3150 (2)	0.0462 (8)
C12	-0.6134 (3)	-0.3220 (3)	0.2506 (2)	0.0449 (8)
C13	-0.5242 (3)	-0.2601 (3)	0.1699 (2)	0.0383 (7)
C14	-0.2884 (3)	-0.3081 (2)	-0.0158 (2)	0.0339 (7)
C15	-0.3860 (3)	-0.3611 (2)	-0.0386 (2)	0.0405 (7)
C16	-0.3696 (4)	-0.4037 (3)	-0.1157 (3)	0.0494 (8)
C17	-0.2562 (4)	-0.3959 (3)	-0.1702 (3)	0.0544 (9)
C18	-0.1588 (4)	-0.3416 (3)	-0.1519 (3)	0.0518 (8)
C19	-0.1752 (3)	-0.2964 (3)	-0.0762 (2)	0.0422 (7)
C20	-0.3479 (3)	-0.1162 (2)	-0.0110 (2)	0.0337 (7)
C21	-0.4568 (3)	-0.0985 (2)	-0.0847 (2)	0.0342 (7)
C22	-0.4565 (3)	-0.0273 (2)	-0.1934 (2)	0.0355 (7)
C23	-0.3471 (2)	0.0400 (2)	-0.2536 (2)	0.0331 (6)
C24	-0.2385 (3)	0.0222 (2)	-0.1855 (2)	0.0348 (7)
C25	-0.2381 (3)	-0.0475 (2)	-0.0764 (2)	0.0362 (7)
C26	-0.3477 (3)	0.1151 (2)	-0.3653 (2)	0.0348 (7)
C27	-0.4634 (3)	0.1464 (2)	-0.4329 (2)	0.0359 (7)
C28	-0.5763 (3)	0.1903 (3)	-0.4052 (3)	0.0478 (8)
C29	-0.6818 (3)	0.2200 (3)	-0.4704 (3)	0.0585 (10)
C30	-0.6758 (3)	0.2055 (3)	-0.5652 (3)	0.0545 (9)
C31	-0.5639 (3)	0.1658 (3)	-0.5959 (3)	0.0477 (8)
C32	-0.4585 (3)	0.1368 (3)	-0.5303 (2)	0.0415 (7)
C33	-0.2325 (3)	0.1731 (2)	-0.4282 (2)	0.0357 (7)
C34	-0.2375 (3)	0.2970 (3)	-0.4935 (2)	0.0410 (7)
C35	-0.1335 (3)	0.3534 (3)	-0.5561 (2)	0.0496 (8)
C36	-0.0240 (3)	0.2880 (3)	-0.5570 (3)	0.0534 (9)
C37	-0.0170 (3)	0.1647 (3)	-0.4944 (3)	0.0541 (9)
C38	-0.1209 (3)	0.1082 (3)	-0.4302 (2)	0.0445 (8)
C15	-0.1602 (8)	0.1440 (8)	0.2444 (8)	0.149 (3)
C25	-0.1369 (9)	0.1050 (6)	0.1644 (15)	0.180 (6)
C35	-0.1713 (6)	0.1783 (6)	0.0572 (8)	0.112 (2)
C45	-0.2240 (5)	0.2821 (5)	0.0336 (4)	0.0844 (13)
C55	-0.2451 (6)	0.3185 (5)	0.1108 (4)	0.101 (2)
C65	-0.2134 (6)	0.2488 (6)	0.2170 (5)	0.107 (2)

Table 2. *Selected geometric parameters* (\AA , $^\circ$)

C1—C14	1.537 (4)	C22—C23	1.463 (4)
C1—C2	1.543 (4)	C23—C26	1.376 (4)
C1—C8	1.544 (4)	C23—C24	1.453 (4)
C1—C20	1.597 (4)	C24—C25	1.337 (4)
C20—C25	1.493 (4)	C26—C33	1.483 (4)
C20—C21	1.503 (4)	C26—C27	1.485 (4)
C21—C22	1.335 (4)		
C14—C1—C2	113.4 (2)	C21—C22—C23	122.8 (3)
C14—C1—C8	111.1 (2)	C26—C23—C24	122.8 (2)
C2—C1—C8	104.2 (2)	C26—C23—C22	122.8 (2)
C14—C1—C20	105.6 (2)	C24—C23—C22	114.4 (2)
C2—C1—C20	110.5 (2)	C25—C24—C23	123.5 (3)
C8—C1—C20	112.1 (2)	C24—C25—C20	124.1 (3)
C25—C20—C21	110.9 (2)	C23—C26—C33	122.6 (2)
C25—C20—C1	113.1 (2)	C23—C26—C27	123.2 (2)
C21—C20—C1	111.6 (2)	C33—C26—C27	114.3 (2)
C22—C21—C20	124.3 (3)		

Program(s) used to solve structure: *SHELXTL-Plus* (Sheldrick, 1987). Program(s) used to refine structure: *SHELXL93* (Sheldrick, 1993).

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: HA1119). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Microbial Transformation of a Dihydroxybiphenyl

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Abstract

The X-ray crystal structure of (3*S*,4*R*)-3,4-dihydroxy-5-(3-hydroxyphenyl)-5-cyclohexen-1-one, C₁₂H₁₂O₄, was determined and showed that the molecule consists of